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United States Patent [19]

Tsumura et al.

[11] **Patent Number:** 5,623,030[45] **Date of Patent:** Apr. 22, 1997[54] **CURABLE COMPOSITION AND PROCESS
FOR PRODUCING MOLDED ARTICLES
USING THE SAME**[75] **Inventors:** Manabu Tsumura; Masafumi
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Kabushiki Kaisha, Osaka, Japan[21] **Appl. No.:** 564,593[22] **Filed:** Nov. 29, 1995[30] **Foreign Application Priority Data**

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[51] **Int. Cl.⁶** **C08F 283/00**[52] **U.S. Cl.** 525/478; 525/479; 528/15;
528/16; 528/17; 528/18; 528/19; 264/331.11[58] **Field of Search** 528/15, 17, 18,
528/19, 16; 525/478, 479; 264/331.11[56] **References Cited****U.S. PATENT DOCUMENTS**5,178,947 1/1993 Charmot et al. 523/212
5,484,867 1/1996 Lichtenhan et al. 528/40*Primary Examiner*—Melvyn I. Marquis
Attorney, Agent, or Firm—Armstrong, Westerman, Hattori,
McLeland & Naughton[57] **ABSTRACT**

A curable composition providing a cured product having a silicon type interpenetrating polymer network, which comprises (A) a silsesquioxane ladder polymer, (B) a silicon compound having at least two SiH groups per molecule, (C) a silicon compound having at least two vinylsilyl groups per molecule and (D) a neutral platinum catalyst; and a process for producing a molded article comprising heating the curable composition under a controlled temperature for synchronously causing hydrolysis and condensation of the alkoxysilyl group and/or silanol group and hydrosilylation.

17 Claims, No Drawings

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CURABLE COMPOSITION AND PROCESS FOR PRODUCING MOLDED ARTICLES USING THE SAME

FIELD OF THE INVENTION

This invention relates to a curable composition and a process for producing a molded article using the same. More particularly it relates to a curable composition comprising a curing system comprised mainly of a silsesquioxane ladder polymer and capable of curing through hydrolysis and condensation of an alkoxysilyl group and/or condensation of a silanol group and a curing system which is capable of curing through hydrosilylation (addition reaction) and comprised of a silicon compound containing at least two SiH groups and/or vinyl groups per molecule and having a number average molecular weight of not more than 1,000 and a catalyst for hydrosilylation, which composition is considered to provide a cured product having a novel silicon type interpenetrating polymer network (hereinafter referred to as IPN); and a process for producing thick-walled articles using the curable composition.

BACKGROUND OF THE INVENTION

JP-A-6-256518 (the term "JP-A" as used herein means an "unexamined published Japanese patent application") (corresponding to U.S. Pat. No. 5,449,734) discloses a curable composition which undergoes condensation reaction and addition reaction in one pot to provide a cured product having a silicon type IPN composed of a network silicon skeleton and a silicon-containing polymer, e.g., polycarbosilane or polysiloxane.

However, the publication shows no guideline for securing the compatibility between a system curing on condensation and a system curing on addition, which seems to be an important prerequisite for IPN formation, and homogeneity of a cured product obtained after curing. Neither does it teach a process for producing a thick-walled molded article having practical mechanical characteristics sufficient for use as heat-resistant structural material.

SUMMARY OF THE INVENTION

An object of the present invention is to provide a curable composition comprising a system curing on condensation (hereinafter referred to as a condensation curing system) and a system curing on addition reaction (hereinafter referred to as an addition curing system) with good compatibility to each other and providing a thick-walled article having excellent mechanical characteristics.

Another object of the present invention is to provide a process for producing a thick-walled molded article which seems to have a novel silicon type IPN and exhibits excellent mechanical characteristics from a curable composition.

The above objects of the present invention are accomplished by composing a curable composition from a condensation curing system which is comprised mainly of a silsesquioxane ladder polymer and an addition curing system (capable of curing through hydrosilylation) which is comprised of silicon compounds having a number average molecular weight of not more than 1,000 and a hydrosilylation catalyst.

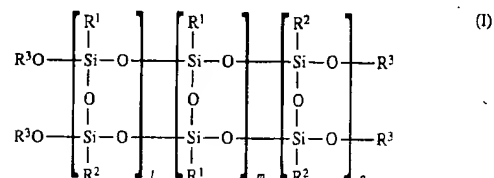
As a result of extensive study on the above-mentioned subject matter, the inventors have found that the curing components used in the present invention, i.e., a condensation curing system and an addition curing system, are

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uniformly compatible with each other before curing over a broad composition ratio, that the composition secures uniformity even after curing, and that a thick-walled cured article free from noticeable cracks can be obtained from the composition by gradually diffusing or volatilizing the organic solvent used and condensation products produced by the condensation reaction. The present invention has been reached based on these findings.

The present invention provides a curable composition comprises:

(A) a silsesquioxane ladder polymer represented by formula (I):



wherein a plurality of R^1 , which may be the same or different, each represent a monovalent hydrocarbon group; a plurality of R^2 , which may be the same or different, each represent a monovalent aromatic hydrocarbon group; R^3 represents a hydrogen atom or a monovalent hydrocarbon group; l , m , and n each represent 0 or a positive integer satisfying $2 \leq l+m+n$, and having a number average molecular weight of not less than 500,

(B) a silicon compound having at least two SiH groups per molecule and having a number average molecular weight of not more than 1,000,

(C) a silicon compound having at least two vinylsilyl groups per molecule and having a number average molecule weight of not more than 1,000, and

(D) a neutral platinum catalyst.

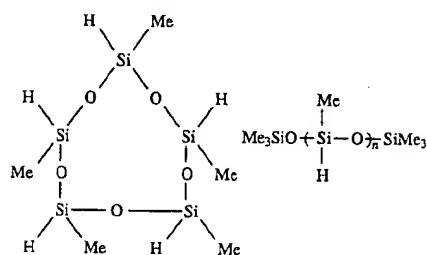
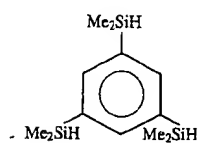
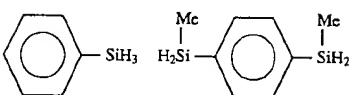
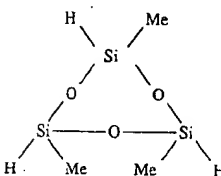
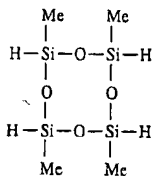
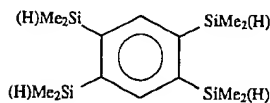
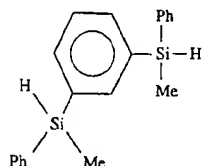
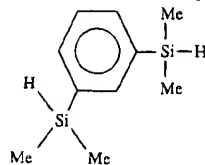
The present invention also relates to a process for producing a molded article, which comprises maintaining the above-mentioned curable composition, as uniformly dissolved or dispersed in 20 to 200 parts by volume of an organic solvent per 100 parts by weight of the silsesquioxane ladder polymer represented by formula (I), at a temperature lower than the boiling point of the organic solvent for 8 hours or more, and then increasing the temperature within a range of from 20° to 400° C., preferably from 20° to 250° C., intermittently or continuously thereby to cause hydrolysis and condensation of the alkoxysilyl group and/or silanol group and hydrosilylation (addition reaction) to proceed synchronously.

DETAILED DESCRIPTION OF THE INVENTION

Component (A) which can be used in the present invention is a component which cures through hydrolysis and condensation. The silsesquioxane ladder polymer as component (A) (hereafter sometimes referred to as silsesquioxane ladder polymer (A)) has a structure represented by formula (I) and a number average molecular weight of not less than 500. Taking into consideration the compatibility with other components constituting the curable composition, and the solubility and reactivity of component (A) itself, the number average molecular weight of component (A) is preferably 500 to 10,000, more preferably 500 to 5,000, most preferably 500 to 1,500.

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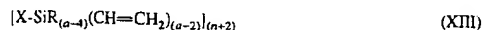
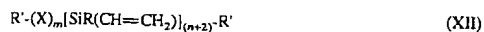
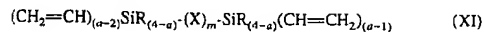
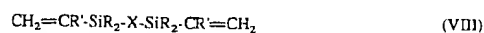


wherein Me is as defined above; Ph represents a phenyl group (hereinafter the same); and n represents an integer of 3 to 5.

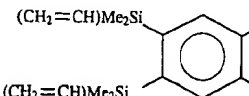
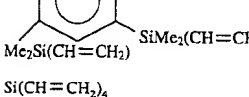
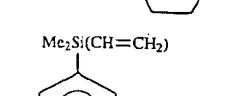
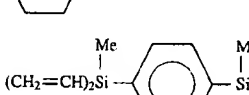
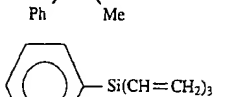
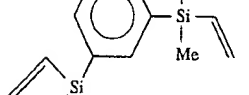
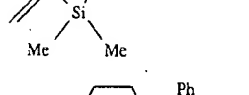
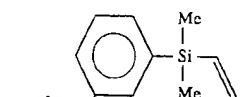
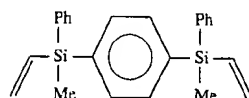
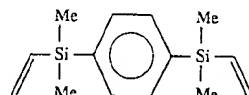
The silicon compound as component (C) is not particularly limited as long as it contains at least two vinylsilyl groups per molecule and has a number average molecular weight of not more than 1,000. Suitable examples of component (C) include vinylsilanes represented by formulae (VIII) to (XIII) shown below and vinylsilanes composed of an aromatic ring with its three or more hydrogen atoms substituted with $\text{SiR}_2(\text{CH}=\text{CH}_2)$, $\text{SiR}(\text{CH}=\text{CH}_2)_2$ or

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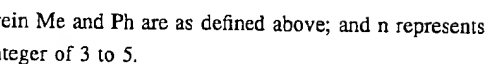
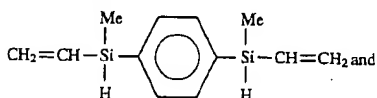
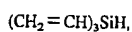
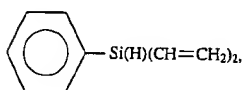
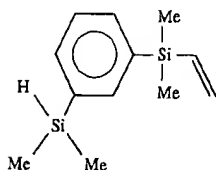
$\text{Si}(\text{CH}=\text{CH}_2)_3$ (wherein R represents a monovalent organic group having 1 to 20 carbon atoms).



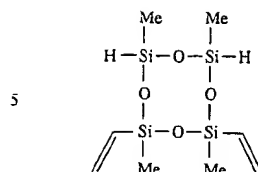
(In the formulae (VIII) to (XIII), R, X, and R' are as defined above for those of formulae (II) to (VII) (preferred structures are also the same); m represents an integer of 1 to 31; and n represents 0 or an integer of 1 to 30.) These silicon compounds may be used either individually or as a combination of two or more thereof. Examples of component (C) which can preferably be used are shown below.



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C[Si](C)(C)c1ccc([Si](C)(C)C=C)cc1

-continued



In the curable composition, a ratio of total SiH groups to total vinylsilyl groups in all the compounds participating in the addition reaction is preferably 0.5 to 5, more preferably 0.6 to 3, and most preferably 0.8 to 2.

The neutral platinum catalyst as component (D) serves as catalyst for hydrosilylation, by which a hydrosilylation reaction, and hydrolysis and condensation reactions of an alkoxy silane proceed in one pot in good yield without being accompanied by production of unfavorable by-products. Such neutral platinum catalysts include platinum-organic compound complexes, platinum-organic functional siloxane complexes, and platinum-diolefin compound complexes. In particular, a platinum-vinylsiloxane complex, a platinum-acetylacetonato complex, and a platinum-decadiene complex are preferred. A recommended, while not limiting, amount of component (D) to be used is 10^{-1} to 10^{-8} mol, particularly 10^{-3} to 10^{-6} mol, per mole of SiH groups.

30 In some cases, the neutral platinum catalyst may be used in combination with a cure retarder to control the reaction rate of hydrosilylation thereby making it possible to syn-
chronously achieve hydrosilylation reaction and alkoxy-
silane hydrolysis and condensation reactions. Useful cure
35 retarders include compounds having an aliphatic unsaturated bond, organophosphorus compounds, organosulfur com-
pounds, nitrogen-containing compounds, tin compounds, and organic peroxides. Examples of the compounds having
an aliphatic unsaturated bond include propargyl alcohol,
40 ene-yne compounds, and maleic esters such as dimethyl
maleate. Examples of the organophosphorus compounds are
triorganophosphines, diorganophosphines, organophospho-
nes, and triorganophosphites. The organosulfur compounds
include organomercaptanes, diorganosulfides, hydrogen sul-
45 fide, benzothiazole, and benzothiazole disulfite. The nitro-
gen-containing compounds include ammonia, primary, sec-
ondary or tertiary alkylamines, arylamines, urea, and
hydrazine. The organic peroxides include di-*t*-butyl perox-
ide, dicumyl peroxide, benzoyl peroxide, and *t*-butyl per-
50 benzoate.

In addition to the essential components (A) to (D), the curable composition of the present invention may further contain (E) a catalyst for silanol condensation, (F) a polyfunctional crosslinking agent capable of silanol condensation, (G) water, and (H) a silica type crosslinking agent.

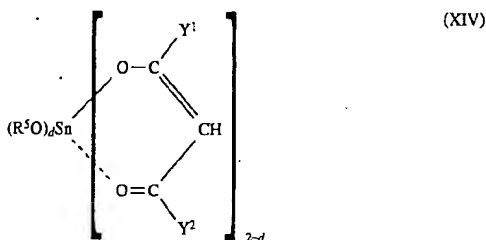
Component (E) serves to accelerate hydrolysis and condensation of the terminal functional groups SiOR^3 of silsesquioxane ladder polymer (A). Where the curable composition contains components (F) and (G) or component (H), component (E) also functions to accelerate the condensation of the alkoxy groups or silanol groups present in compounds (F) or (H) or silanol groups produced by hydrolysis or to accelerate the condensation between these functional groups and the terminal functional groups SiOR^3 of silsesquioxane ladder polymer (oligomer) (A).

The catalyst for silanol condensation as component (E) is selected from a broad range of known catalysts, such as

various acid catalysts, alkali catalysts and organometallic compounds. Specific but non-limiting examples of suitable catalysts (E) are as follows. Suitable acid catalysts include hydrochloric acid, sulfuric acid, nitric acid, acetic acid, phosphoric acid, phosphoric esters, active clay, iron chloride, boric acid, trifluoroacetic acid, trifluoromethanesulfonic acid, and p-toluenesulfonic acid. Suitable alkali catalysts are hydroxides of alkali metals or alkaline earth metals, alkoxides of alkali metals or alkaline earth metals, tetraalkylammonium hydroxides, tetraalkylphosphonium hydroxides, and amine compounds. Specific examples of the amine compounds include pyridine, picoline, lutidine, pyrazine, piperidine, piperazine, piperazine, pyrazole, pyridazine, pyrimidine, pyrrolidine, butylamine, octylamine, laurylamine, dibutylamine, monoethanolamine, triethylenetetramine, oleylamine, cyclohexylamine, benzylamine, diethylaminopropylamine, xylylenediamine, triethylenediamine, guanidine, diphenylguanidine, 2,4,6-tris(dimethylaminomethyl)phenol, morpholine, N-methylmorpholine, 2-ethyl-4-methylimidazole, 1,8-diazabicyclo[5,4,0]undecene (DBU), carboxylic acid salts of these amine compounds, low-molecular weight polyamide resins obtained from an excess polyamine and a polybasic acid, the reaction product between an excess polyamine and an epoxy compound, and amino-containing silane coupling agents, such as γ -aminopropyltrimethoxysilane and N-(β -aminoethyl)aminopropylmethyldimethoxysilane. Fluorine compounds, such as tetrabutylammonium fluoride, potassium fluoride and sodium fluoride, are also useful.

The organometallic compounds as component (E) include organic acid salts, alkoxides and chelates of tin, lead, zinc, iron, cobalt, titanium, aluminum, zirconium, boron, etc.

Examples of suitable organotin catalysts include those represented by formula (XIV):



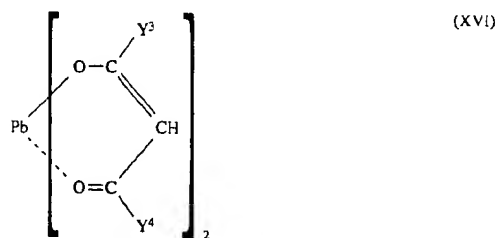
wherein R^5 represents a substituted or unsubstituted monovalent hydrocarbon group, such as an alkyl group, a substituted alkyl group (e.g., chloromethyl), an alkenyl group (e.g., vinyl or allyl) or an aryl group (e.g., phenyl or tolyl); Y^1 and Y^2 each represent an alkyl or alkoxy group having 1 to 8 carbon atoms; and d represents 0, 1 or 2, and those represented by formula (XV):



wherein R^6 represents a substituted or unsubstituted monovalent hydrocarbon group, such as an alkyl group, a substituted alkyl group (e.g., chloromethyl), an alkenyl group (e.g., vinyl or allyl) or an aryl group (e.g., phenyl or tolyl).

Specific examples of these organotin compounds are tin (II) methoxide, tin (II) ethoxide, tin (II) 2,4-pentanedionate, tin (II) octanoate, tin (II) acetate, dibutyltin dilaurate, dibutyltin malate, dibutyltin diacetate, tin naphthenate, a reaction product between dibutyltin oxide and a phthalic ester, and dibutyltin diacetylacetonate.

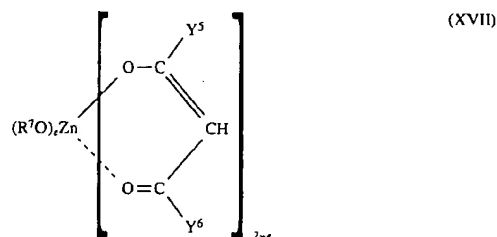
The organolead catalysts include those represented by formula (XVI):



wherein Y^3 and Y^4 each represent a substituted or unsubstituted alkyl or alkoxy group having 1 to 8 carbon atoms.

Specific examples of the compounds of formula (XVI) are lead (II) hexafluoropentanedionate, lead (II) 2,4-pentanedionate, lead (II) 2,2,6,6-tetramethyl-3,5-heptanedionate, and lead octylate.

The organozinc catalysts include those represented by formula (XVII):



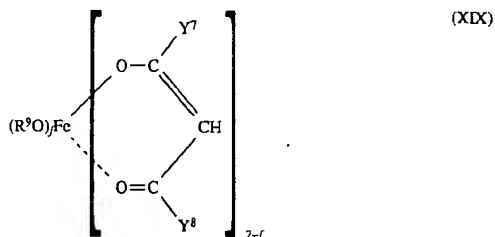
wherein R^7 represents a substituted or unsubstituted monovalent hydrocarbon group, such as an alkyl group, a substituted alkyl group (e.g., chloromethyl), an alkenyl group (e.g., vinyl or allyl) or an aryl group (e.g., phenyl or tolyl); Y^5 and Y^6 each represent an alkyl or alkoxy group having 1 to 8 carbon atoms; and e represents 0, 1 or 2, and those represented by formula (XVIII):



wherein R^8 represents a substituted or unsubstituted monovalent hydrocarbon group, such as an alkyl group, a substituted alkyl group (e.g., chloromethyl), an alkenyl group (e.g., vinyl or allyl) or an aryl group (e.g., phenyl or tolyl).

Specific examples of these organozinc compounds are zinc dimethoxide, zinc diethoxide, zinc methoxyethoxide, zinc 2,4-pentanedionate, zinc acetate, zinc 2-ethylhexanoate, zinc formate, zinc methacrylate, zinc neodecanoate, zinc undecylenate, and zinc octylate.

The organoiron catalysts include those represented by formula (XIX):



wherein R^9 represents a substituted or unsubstituted monovalent hydrocarbon group, such as an alkyl group, a substituted alkyl group (e.g., chloromethyl), an alkenyl group (e.g., vinyl or allyl) or an aryl group (e.g., phenyl or tolyl).

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preferably 0.5 to 6 parts by weight, per 100 parts by weight of silsesquioxane ladder polymer (A).

The polyfunctional silicon compound capable of silanol condensation as component (F) preferably includes those represented by formula (XXVII):



wherein R^4 represents a monovalent organic group; and k represents a positive integer satisfying $1 \leq k \leq 7$. Specific examples of the silicon compound of formula (XXVII) are $\text{Si}(\text{OEt})_4$, $\text{MeSi}(\text{OMe})_3$, $\text{Si}(\text{OAc})_4$, $\text{MeO}\{\text{Si}(\text{OMe})_2\text{O}\}_n\text{Me}$ ($n=3$ to 6 in average), and $\text{EtO}\{\text{Si}(\text{OEt})_2\text{O}\}_n\text{Et}$ ($n=3$ to 6 in average), wherein Et is an ethyl group; and Ac is an acetyl group (hereinafter the same). Additional examples of useful polyfunctional silicon compounds include $\text{MeSi}(\text{OMe})_2$, $\text{MeSi}(\text{OEt})_3$, $\text{MeSi}(\text{OAc})_3$, $\text{Ph}_2\text{Si}(\text{OH})_2$, PhMe_2SiOH , Ph_2MeSiOH , $\text{PhSi}(\text{OH})_3$ or a low-molecular weight oligomer thereof, Me_3SiOH , and $\text{MeSi}(\text{OH})_3$ or a low-molecular weight oligomer thereof.

Component (F) is used in an amount of 5 to 50 parts by weight, preferably 10 to 50 parts by weight, per 100 parts by weight of silsesquioxane ladder polymer (A). If the amount of component (F) added is less than 5 parts by weight, the effect of addition is insubstantial. If it exceeds 50 parts, a cured product of the curable composition tends to be brittle.

Water as component (G) can be added in order to accelerate the hydrolysis of the alkoxysilyl groups present in the curable composition thereby to accelerate the condensation reaction sufficiently. Water is added in an amount of 10 to 100 mol % based on the total alkoxy group content in the curable composition. If the amount of water added exceeds 100 mol %, the curable composition tends to fail to provide a uniform cured product.

The silica type crosslinking agent which can be used as component (H) includes fine powder of silica hydrate or anhydrous silica or silica powder having been treated with various surface treating agents. The silanol group of the silica powder or water adsorbed on the silica powder participates in the condensation of the terminal functional groups SiOR^3 of silsesquioxane ladder polymer (A) thereby bringing about improvement in physical properties of the resulting cured product. Component (H) is preferably used in an amount of 5 to 30 parts by weight per 100 parts by weight of silsesquioxane ladder polymer (A).

The composition ratio of the condensation curing system (component (A)) and the addition curing system (components (B) to (D)) in the curable composition is subject to variation depending on the use of the curable composition and the cured product thereof. For example, where the cured product is intended for use as structural material demanded to have high elasticity and high strength, the condensation curing system/addition curing system ratio is preferably 1 or more, still preferably 2 or more, most preferably 3 or more, by weight. Where use as rubbery material is intended, the ratio is preferably not more than 1, still preferably not more than $\frac{1}{2}$, particularly preferably not more than $\frac{1}{3}$, by weight.

In the present invention, an organic solvent can be used for uniformly mixing the above-described components. It is desirable for the organic solvent to be capable of sufficiently dissolving silsesquioxane ladder polymer (A) and capable of dissolving water to some extent. Such organic solvents include hydrocarbons, such as benzene and toluene; ethers, such as tetrahydrofuran, 1,4-dioxane, and diethyl ether; ketones, such as acetone and methyl ethyl ketone; esters,

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such as ethyl acetate; and halogenated hydrocarbons, such as chloroform, methylene chloride, and 1,2-dichloroethane. These solvents may be used either individually or as a mixture of two or more thereof. Tetrahydrofuran and chloroform are preferred solvents.

The organic solvent is used in an amount of 20 to 200 ml, preferably 60 to 120 ml, per 100 g of silsesquioxane ladder polymer (A). If used in amounts less than 20 ml, it is difficult to solve silsesquioxane ladder polymer (A). If the amount exceeds 200 ml, a cured product free from air bubbles and cracks is hard to obtain.

The process for producing a molded article from the curable composition of the present invention will be described below. A molded article free from noticeable cracks can be produced from the curable composition of the present invention by heating the composition to cure while controlling the rate of temperature increase so as to realize a good balance of a rate of condensation curing, a rate of volatilization of the volatile matter from the curing system, and a rate of diffusion of the volatile matter remaining in the system. In carrying out curing, the curable composition is first maintained at a temperature lower than the boiling point of the organic solvent used, e.g., about 20° to 50°C ., for at least 8 hours and then the temperature is increased within a range of from 20° to 400°C ., preferably 20° to 250°C ., either intermittently or continuously. A thick-walled article can be obtained by, for example, casting the curable composition into a mold, the inner wall of which has thereon a polyimide film via a double-sided adhesive tape, placing the mold with a cap on horizontally in a hot air drier, and gradually elevating the temperature to cause cure. The polyimide film adhered to the inner wall of the mold provides easy release for a composition being cured so that a cured product may easily release from the mold even if it shrinks and thereby be prevented from initiating cracks. Heat cure is conducted by increasing the temperature within a range of from 20° to 400°C . either intermittently or continuously. In the case of continuously increasing temperature, gradual increase at a rate of not more than 5°C/hr is preferred. In the case of intermittent temperature increase, heating with successive, at 50°C . for 8 to 24 hours, at 80°C . for 8 to 24 hours, at 100°C . for 8 to 24 hours, and then at 150°C . for 12 to 70 hours may be mentioned as a preferred example of temperature increase.

If desired, a product after heat cure may further be subjected to heat treatment in order to meet the requirement of improvement in physical properties or the end use. The heat treatment can be carried out at 150° to 450°C . in air or an inert gas, such as nitrogen or argon, or under reduced pressure.

The present invention will now be illustrated in greater detail by way of Examples, but it should be understood that the present invention is not deemed to be limited thereto. Unless otherwise indicated, all the percents and parts are given by weight.

In Examples, the following commercially available products were used.

1) Silsesquioxane ladder polymer (A):

"Glass Resin (GR)" series produced by Owens Illinois Incorporated and purchased from Showa Denko K. K. were used.

1-1) GR-100 (Ph/Me on silicon: 1/2; weight average molecular weight (Mw)/number average molecular weight (Mn): 7210/1260 on polystyrene standard conversion by gel-permeation chromatography (GPC) (hereinafter the same))

1-2) GR-950 (Ph/Me on silicon: 1/0; Mw/Mn: 1080/690)

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- 1-3) 3/1 (by weight) mixture (GR-150*; Ph/Me on silicon: 1/1) of GR-100 and GR-908 (Ph/Me on silicon: 4/1; Mw/Mn: 1270/680)
- 2) Polyfunctional crosslinking agent (F):
"Methyl Silicate 51", a compound of formula (XXVII) wherein $R^4=Me$ and $k=4$ (average), produced by Colcoat Company Ltd.
- 3) Silica type crosslinking agent:
Silica hydrate "NIPSIL-LP" produced by Nippon Silica Kogyo K. K.

EXAMPLE 1

In a 50 ml tube were put 5.0 g of GR-150* and 2.5 g of Methyl Silicate 51. To the mixture was added 5 cc of tetrahydrofuran (THF) as a solvent, and Glass Resin was completely dissolved by means of an ultrasonic cleaning system. A condensation catalyst $Ti(O-iPr)_2(acac)_2$ (500 mg) was added to the solution. In a separate 10 ml tube were put 918 mg (3.5 mmol) of 1,4-bis(dimethylvinylsilyl)benzene, 449 mg (2.3 mmol) of 1,4-bis(dimethylsilyl)benzene, and 211 mg (0.88 mmol) of 1,3,5,7-tetramethylcyclotetrasiloxane, and 30 mg of a 10% THF solution of dimethyl maleate (cure retarder) was added thereto, followed by shaking lightly to mix (300 equivalents to a platinum catalyst hereinafter described). The carbosilane solution and the above prepared silsesquioxane ladder polymer solution were mixed together, and 7.2 mg of a Pt-vinylsiloxane complex (synthesized from $H_2PtCl_6 \cdot 6H_2O$ [Me₂(CH₂=CH)Si]₂O/NaHCO₃/toluene; 9.71×10^{-6} mmol/mg; 1×10^{-6} equivalent to the Si-vinyl group).

A 25 μ m thick polyimide film was laid on the inner wall of an ointment can of 6.7 cm in diameter via a double-sided adhesive tape, and the above prepared mixed solution was slowly cast in the can. The can was placed horizontally in a hot air drier, the cap was fitted on, and the can and the contents were heated at 50° C. for 18 hours, at 80° C. for 9 hours, at 100° C. for 14 hours, and then at 150° C. for 22 hours for curing. The resulting cured product was a 2.3 mm thick clear brown solid. A test specimen of about 40 mm in length and about 5 mm in width was cut out of the solid with a diamond cutter.

The amounts of condensation curing system components and the addition curing system components used in the curing composition are shown in Tables 1 and 2 shown below, respectively.

EXAMPLES 2 TO 17

A curable composition was prepared in the same manner as in Example 1 except for the alterations shown in Tables 1 and 2 below. A cured product was prepared and a test specimen was cut out therefrom in the same manner as in Example 1, provided that in Examples in which the Ti catalyst was not used, cure was conducted by heating at 50° C. for 18 hours, at 80° C. for 9 hours, at 100° C. for 40 hours, and then at 150° C. for 72 hours.

The cured products and test specimens prepared in Examples 1 to 17 were evaluated in terms of flexural characteristics, thermogravimetric loss, and oxygen index in accordance with the following test methods.

1) Flexural Characteristics:

Flexural modulus, flexural strength, and flexural elongation were measured by means of a precision universal tester produced by Shimadzu Corporation in accordance with JIS K7203 (flexural test method using small specimens; span: 15 mm; indenter: 5R; support: 2R; test speed: 0.5 mm/min).

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2) Thermogravimetric Loss:

A small piece of about 10 mg was precisely weighed out of the cured product and thermogravimetrically analyzed by means of TGA-50 manufactured by Shimadzu Corp. The piece was heated in a nitrogen stream (30 ml/min) at a rate of 20° C./min, and the temperature at 5% weight loss (Td₅) was measured.

3) Limiting Oxygen Index (LOI):

Limiting oxygen index was determined using Suga Tester Model ON-1 in accordance with JIS 7201. Because the test specimen was small, wire was coiled around the lower part of the specimen, and the specimen was fixed about 10 cm down the upper part of the combustion cylinder.

The oxygen and nitrogen flow rates were adjusted with valves, and a gas stream having a prescribed oxygen concentration was made to flow from the bottom for at least 30 seconds. The upper part of the specimen was ignited with an ignition source, and the condition of combustion was observed. When the upper part of a specimen turned red but went out within 3 minutes (mostly within 20 seconds), the LOI was judged to be higher than the oxygen concentration of the gas, while the LOI of a specimen which continued burning more than 3 minutes was judged to be lower than the oxygen concentration. The higher the LOI, the higher the flame retardancy.

The results of the evaluation are shown in Table 3 below. The amounts of condensation curing system components and the addition curing system components used in the curing composition are shown in Tables 1 and 2 below, respectively.

TABLE 1

Condensation Curing System of Curing Composition					
Example No.	Ph/Me Ratio in (A)	Catalyst (E) (part)	Methyl Silicate (F) (part)	H ₂ O (G) (part)	Silica* (H) (part)
1	1/1	10	50	—	—
2	1/1	10	30	—	—
3	1/2	3	—	—	—
4	1/2	3	30	2	—
5	1/2	3	30	—	—
6	1/2	3	30	6	—
7	1/2	—	30	—	—
8	1/2	—	—	—	10
9	1/2	3	—	—	10
10	1/2	—	30	—	10
11	1/2	—	—	4	10
12	1/2	—	30	4	—
13	1/2	3	30	—	10
14	1/2	3	—	4	10
15	1/2	—	30	4	10
16	1/2	3	—	4	—
17	1/2	3	—	4	—

Note: *NIPSIL-LP (produced by Nippon Silica Kogyo K.K.)

TABLE 2

Addition Curing System of Curing Composition				
Example No.	SiH Component ¹⁾ (B)	Si-vinyl Component ²⁾ (C)	SiH/Si-vinyl Molar Ratio	Dimethyl Maleate Equiv. ³⁾
1	a + b	c	1.16	300
2	b	c	1.14	30
3	b	d	1.17	30
4	b	d	1.17	30
5	b	c	1.17	100
6	b	c	1.17	100

TABLE 2-continued

Addition Curing System of Curing Composition				
Example No.	SiH Component ¹⁾ (B)	Si-vinyl Component ²⁾ (C)	SiH/Si-vinyl Molar Ratio	Dimethyl Maleate Equiv. ³⁾
7	b	c	1.00	100
8	b	c	1.00	100
9	b	c	1.00	100
10	b	c	1.00	100
11	b	c	1.00	100
12	b	c	1.00	100
13	b	c	1.00	100
14	b	c	1.00	100
15	b	c	1.00	100
16	b	c	1.00	500
17	b	c	1.00	500

Note:

¹⁾SiH Component

a: 1,4-Bis(dimethylsilyl)benzene

b: 1,3,5,7-Tetramethylcyclotetrasiloxane

²⁾Si-vinyl Component

c: 1,4-Bis(dimethylvinylsilyl)benzene

d: $[\text{CH}_2(\text{CH}_2=\text{CH})\text{SiO}]_4$ ³⁾Equivalent to Pt-vinylsiloxane (used in an amount of 1×10^{-5} equiv. to Si-vinyl group); dimethyl maleate was used as a cure retarder.

In 100 ml of dried toluene were dissolved 18.02 mg (1.03 mmol) of methylphenyldivinylsilane and 226 μl of a 1 wt % toluene solution of a platinum-vinylsiloxane complex (1.88×10^{-4} mmol; 1×10^{-4} mmol per mmole of SiH group) (synthesized from $\text{H}_2\text{PtCl}_6 \cdot 6\text{H}_2\text{O}/[\text{Me}_2(\text{CH}_2=\text{CH})\text{Si}]_2\text{O}/\text{NaHCO}_3/\text{toluene}$). To the Pt catalyst solution was slowly added dropwise 100 ml of a solution of 18.30 g (94 mmol) of 1,4-bis(dimethylsilyl)benzene in dried toluene at room temperature in a nitrogen atmosphere. After the addition, the reaction mixture was stirred at room temperature for one day while evaporating the volatile matter to obtain a crude polymer. The crude polymer was dissolved in hexane and subjected to column chromatography on silica gel to collect 9.68 g of reactive silicone polymer fraction 1 represented by formula: $\text{CH}_2=\text{CH}[\text{Si}(\text{Ph})(\text{Me})\text{CH}_2\text{CH}_2\text{SiMe}_2\text{C}_6\text{H}_4\text{-p-SiMe}_2\text{CH}_2\text{CH}_2\text{Si}(\text{Ph})(\text{Me})\text{CH}=\text{CH}_2$ (Me is a methyl group and Ph is a phenyl group) as a hexane fraction and 12.60 g of fraction 2 as a toluene fraction. Fractions 1 and 2 had a molecular weight distribution Mw/Mn of 3400/1700 and 8700/5700, respectively, as measured by GPC and a vinyl content of 81.3 mmol/100 g and 29.6 mmol/100 g, respectively.

Reactive silicone polymer fraction 1 (500 mg), 4.9 μl of a platinum catalyst solution (synthesized from $\text{H}_2\text{PtCl}_6 \cdot 6\text{H}_2\text{O}/[\text{Me}_2(\text{CH}_2=\text{CH})\text{SiO}]_4/\text{NaHCO}_3/\text{toluene}$;

TABLE 3

Physical Properties of Cured Product											
Example No.	Composition Ratio*	Gel Fraction (%)	Thickness (mm)	Appearance	Flexural Characteristics			TG Analysis			
					Modulus (GPa)	Strength (MPa)	Elongation (%)	Weight Loss		Td ₅ (°C.)	LOI
1	8/2	97	2.0	clear brown	1.94	34.2	1.9	3.2	6.2	469	—
2	"	98	1.9	"	1.91	37.3	2.1	2.0	3.6	544	28
3	"	100	2.2	"	1.84	43.2	3.1	0.5	1.6	>580	—
4	"	95	1.8	clear yellow	1.71	33.0	2.0	1.3	2.2	>580	—
5	"	95	2.0	"	2.16	39.6	1.9	1.8	3.0	563	—
6	"	96	1.8	"	2.27	41.6	2.0	2.3	3.6	549	32
7	"	95	1.2	semitransparent white	1.82	33.4	1.5	2.2	3.0	530	34
8	"	100	1.0	semitransparent white	2.21	43.2	3.0	0.8	2.0	>580	38
9	"	98	1.1	semitransparent white	1.90	31.0	1.6	1.8	3.4	520	—
10	"	96	1.4	semitransparent yellow	1.80	30.8	2.2	1.4	2.6	560	35
11	"	99	1.3	semitransparent white	2.02	44.0	2.3	1.2	2.5	>580	39
12	"	100	1.5	semitransparent white	1.54	31.4	1.5	2.8	4.5	509	—
13	"	97	1.4	clear yellow	2.24	40.8	3.4	3.4	6.9	454	36
14	"	98	1.3	"	2.10	36.0	2.3	0.4	1.4	>580	38
15	"	96	1.4	semitransparent white	1.74	31.0	1.8	3.1	5.8	481	33
16	5/5	98	2.7	clear yellow	1.20	40.0	3.4	0.8	3.1	530	30
17	2/8	98	1.3	"	1.78	78.0	10.7	0.7	2.9	541	28

Note: *Weight ratio of condensation curing system/addition curing system in the curable composition.

It is seen from the results in Table 3 that the curable 8.3×10^{-7} mmol/ μl solution; 1×10^{-5} mmol per SiH group),

The resulting cured product was found heterogeneous, consisting of a soft portion and a glassy portion. The soft portion had a thermogravimetric loss as large as 28.5% at 500° C., and that of the glassy portion was 19.0%, still higher than that of the cured product obtained in Examples. Further, observation under a phase-contrast microscope revealed large domains of 50 μ m or more.

Comparative Example 2

1,1,3,3,5,5-Hexamethyltrisiloxane (7.5 mmol), 7.5 mmol of divinylmethylphenylsilane, 1.25 mmol of 1,3,5,7-tetramethylcyclotetrasiloxane, and 3.2 g of Ethyl Silicate 40 (produced by Colcoat Company Ltd.) were mixed, and 50 mmol of water, 2×10^{-3} mmol of a platinum-vinylsiloxane complex, 1 mmol of di(2-ethylhexyl) phosphite, and 5 ml of isopropyl alcohol were added thereto, followed by stirring well to prepare a uniform clear solution.

The resulting solution was cast in the same ointment can as used in Example 1 and cured under the same heating condition as in Example 1. The resulting cured product suffered cracks all over the surface and was non-uniform due to undercure in the inside thereof. The physical properties of the cured product were unmeasurable.

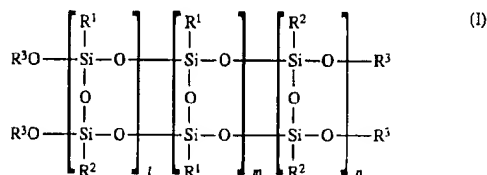
As described above, the present invention provides a curable composition which provides on curing a cured product having a silicon type IPN composed of a network silicon skeleton and a silicon-containing polymer, such as polycarbosilane or polysiloxane, in which compatibility between a condensation curing system and an addition curing system before curing and homogeneity after curing can be secured. The cured product obtained from the curable composition exhibits excellent heat resistance and mechanical characteristics and are therefore useful in aerospace industries; transportation means such as automobiles, railroads, vessels, etc., which are expected to increase the speed in the future; and large-sized structures, such as high rises, deep subterranean structures, and submarine structures.

While the invention has been described in detail and with reference to specific embodiments thereof, it will be apparent to one skilled in the art that various changes and modifications can be made therein without departing from the spirit and scope thereof.

What is claimed is:

1. A curable composition comprising:

(A) a silsesquioxane ladder polymer represented by formula (I):



wherein a plurality of R^1 , which may be the same or different, each represent a monovalent hydrocarbon group; a plurality of R^2 , which may be the same or different, each represent a monovalent aromatic hydrocarbon group; R^3 represents a hydrogen atom or a monovalent hydrocarbon group; l, m, and n each represent 0 or a positive integer satisfying $2 \leq l+m+n$, and having a number average molecular weight of not less than 500,

(B) a silicon compound having at least two SiH groups per molecule and having a number average molecular weight of not more than 1,000,

(C) a silicon compound having at least two vinylsilyl groups per molecule and having a number average molecular weight of not more than 1,000, and

(D) a neutral platinum catalyst.

2. A curable composition according to claim 1, wherein the composition further comprises (E) a catalyst for silanol condensation.

3. A curable composition according to claim 1, wherein the composition further comprises (F) a polyfunctional crosslinking agent capable of silanol condensation of the terminal functional groups of the silsesquioxane ladder polymer.

4. A curable composition according to claim 1, wherein the composition further comprises (G) water.

5. A curable composition according to claim 1, wherein the composition further comprises (H) a silica crosslinking agent selected from the group consisting of silica hydrate, anhydrous silica and silica having been treated with surface treating agents.

6. A curable composition according to claim 1, wherein the composition further comprises (E) a catalyst for silanol condensation and (F) a polyfunctional crosslinking agent capable of silanol condensation of the terminal functional groups of the silsesquioxane ladder polymer.

7. A curable composition according to claim 1, wherein the composition further comprises (E) a catalyst for silanol condensation and (G) water.

8. A curable composition according to claim 1, wherein the composition further comprises (E) a catalyst for silanol condensation and (H) a silica crosslinking agent selected from the group consisting of silica hydrate, anhydrous silica and silica having been treated with surface treating agents.

9. A curable composition according to claim 1, wherein the composition further comprises (F) a polyfunctional crosslinking agent capable of silanol condensation of the terminal functional groups of the silsesquioxane ladder polymer and (H) a silica crosslinking agent selected from the group consisting of silica hydrate, anhydrous silica and silica having been treated with surface treating agents.

10. A curable composition according to claim 1, wherein the composition further comprises (G) water and (H) a silica crosslinking agent selected from the group consisting of silica hydrate, anhydrous silica and silica having been treated with surface treating agents.

11. A curable composition according to claim 1, wherein the composition further comprises (F) a polyfunctional crosslinking agent capable of silanol condensation of the terminal functional groups of the silsesquioxane ladder polymer and (G) water.

12. A curable composition according to claim 1, wherein the composition further comprises (E) a catalyst for silanol condensation and (F) a polyfunctional crosslinking agent capable of silanol condensation of the terminal functional groups of the silsesquioxane ladder polymer and (G) water.

13. A curable composition according to claim 1, wherein the composition further comprises (E) a catalyst for silanol condensation and (F) a polyfunctional crosslinking agent capable of silanol condensation of the terminal functional groups of the silsesquioxane ladder polymer and (H) a silica crosslinking agent selected from the group consisting of silica hydrate, anhydrous silica and silica having been treated with surface treating agents.

14. A curable composition according to claim 1, wherein the composition further comprises (E) a catalyst for silanol condensation, (G) water and (H) a silica crosslinking agent selected from the group consisting of silica hydrate, anhydrous silica and silica having been treated with surface treating agents.

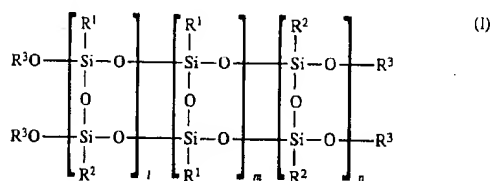
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15. A curable composition according to claim 1, wherein the composition further comprises (F) a polyfunctional crosslinking agent capable of silanol condensation of the terminal functional group of the silsesquioxane ladder polymer, (G) water and (H) a silica crosslinking agent selected from the group consisting of silica hydrate, anhydrous silica and silica having been treated with surface treating agents.

16. A curable composition according to claim 1, wherein the composition further comprises (E) a catalyst for silanol condensation, (F) a polyfunctional crosslinking agent capable of silanol condensation of the terminal functional groups of the silsesquioxane ladder polymer, (G) water and (H) a silica crosslinking agent selected from the group consisting of silica hydrate, anhydrous silica and silica having been treated with surface treating agents.

17. A process for producing a molded article, which comprises the steps of: uniformly dissolving or dispersing a curable composition comprising:

(A) a silsesquioxane ladder polymer represented by formula (I):



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wherein a plurality of R^1 , which may be the same or difference, each represent a monovalent hydrocarbon group; a plurality of R^2 , which may be the same or difference, each represent a monovalent aromatic hydrocarbon group; R^3 represents a hydrogen atom or a monovalent hydrocarbon group; l , m and n each represent 0 or a positive integer satisfying $2 \geq l+m+n$, and having a number average molecular weight of not less than 500,

(B) a silicon compound having at least two SiH groups per molecule and having a number average molecular weight of not more than 1,000,

(C) a silicon compound having at least two vinylsilyl groups per molecule and having a number average molecular weight of not more than 1,000, and

(D) a neutral platinum catalyst, in 20 to 200 parts by volume of an organic solvent per 100 parts by weight of said silsesquioxane ladder polymer, and maintaining said composition at a temperature lower than the boiling point of said organic solvent for 8 hours or more, casting the composition into a mold and then intermittently or continuously increasing the temperature within a range from 20° to 400° C. thereby to cause hydrolysis and condensation of the alkoxysilyl group and/or silanol group and hydrosilylation to proceed synchronously so as to cure the composition.

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